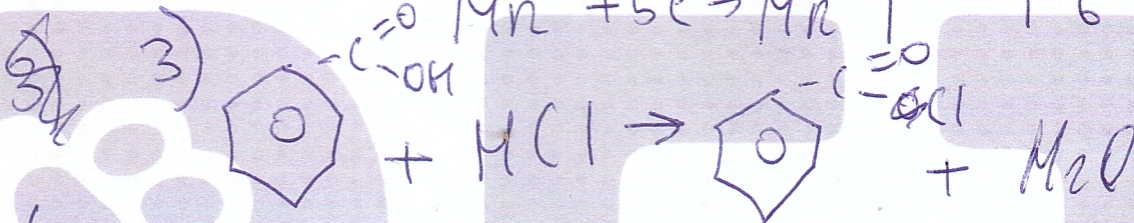
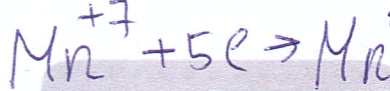
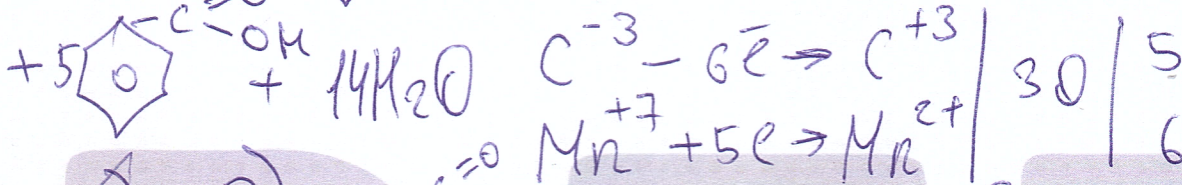
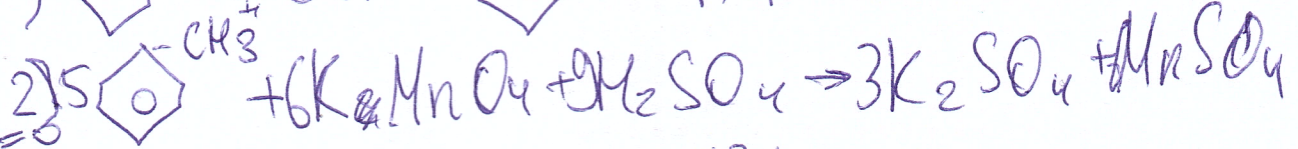
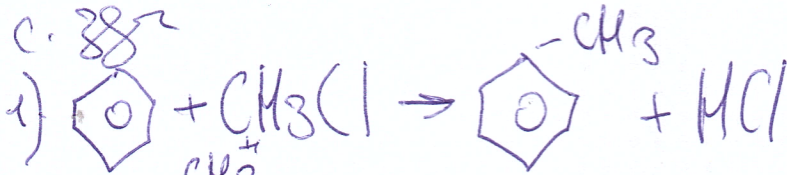


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Подпись

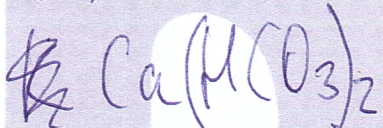
N6) с. 88



N5

металл Cr (хром)

N1



$$M(\text{Ca}(\text{HCO}_3)_2) = 162 \text{ г/моль.}$$

$$n = \frac{0,8 \text{ г}}{22,4} = 0,03 \text{ моль}$$

(n·M)

$$m(\text{Cr-раств}) = 0,03 \times 162 = 4,86 \text{ г.}$$

Ответ: 4,86 г.

N4

$$133,9 \text{ г/моль}$$

$$20\% = 107,1 \text{ г/моль}$$

$$20\% = 26,78 \text{ г/моль}$$

$$m(\text{сб-ла}) = m = 63 \times 6 = 378$$

$$m = 126 \times 6 = 756$$

$$\frac{133,9}{22,4} = 6 \text{ моль}$$

$$\frac{1000}{63} = 15,8$$

$$\frac{1000}{126} = 7,9$$

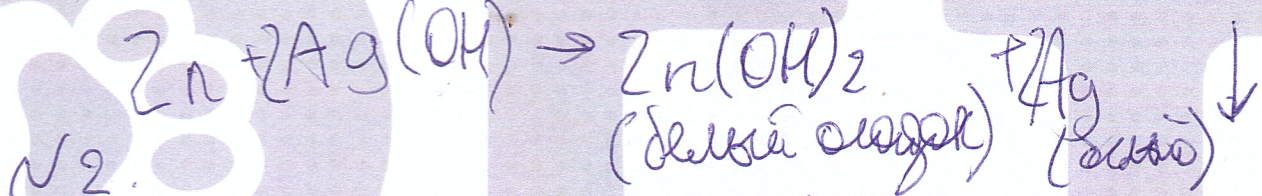
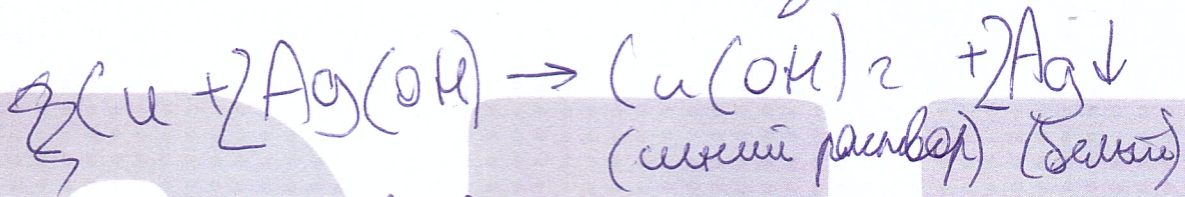
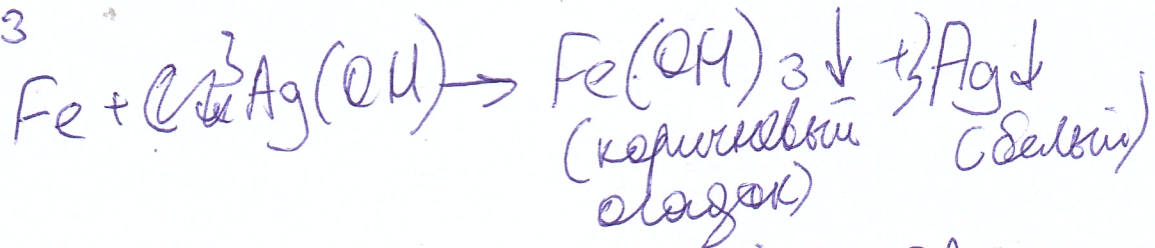
$$+ = 1134$$

Дата 02.04.2026

№ участника 246-3

Подпись 

№3



№2

